

كلية العلوم قسم النبات المنصورة - مصر

Final Examination in Botany First Term: Jan. 2016

	<u>L'enleur</u>			
Educational	Year:	Second	level	

Code: B 202

Program (Branch): Biology Students Courses: Introduction in plant metabolism

Time: 2 hrs

Date: 27 /12 /2015

Full mark: 60

Answer	the	foll	owing	aues	stions:	
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Section I:Photosynthesis

Answer the following:

- A- Can you discuss briefly what exactly happens during the 1st stage of photosynthesis? (3 marks)
- B- From the 2nd phase of photosynthesis:
 - 1- Where does it occur? (0.5 mark)
 - 2- What is the name of this phase? (0.5 mark)
 - 3- What is the CO₂ acceptor and what is the responsible enzyme? (1 mark)
 - 4- Mention the steps which need NADPH2 and ATP. (2 marks)
 - 5- Do you think that RuBP can be regenerated? (0.5 mark)
 - 6- Give an equation to summarize this cycle? Comment on this equation. (2 marks)
 - 7- What are the different enzymes required until this phase is completed? Mention the function of two enzymes. (2 marks)
- C- Show with an equation how can higher plants synthesize source? (1.5 marks)

Section II: Respiration

A- Complete the following sentences: (3 marks; 1 mark each)

- 2. Conversion of 3-phopsphoglycerate into 2-phosphoglycerate is catalyzed by
- 3. For each ATP molecule formed, H+ is transferred from to

B- Answer the following:

- 1. Illustrate the different steps of conversion of glucose (6 C) into two 3C molecules. (4 marks)
- 2. Starting with pyruvate, write all the reactions needed to convert it into CO₂. (3 marks)

C- Choose the most correct answer. (3 Marks; 1 mark each)

- 1. Six electrons from NADH+H⁺ and four electrons from FADH₂ moved through ETC form molecules of ATP.
 - a-3 & 2

b-6&4

c-9&4

d- none of the previous

- 2. All of these compounds contain iron sulfur (Fe-S) centers except
 - a- Complex I

b- Complex II

c- Complex III

d- Complex IV

- 3. Cytochrome C is a carrier protein characterized by
 - a- Hydrophilic

b- Peripherally attached

c- Hydrophobic

d- all the previous

Section III: Carbohydrates & Lipids

A- Write the right concept for the following sentences: (3.5 marks)

1. (.....) are small saccharide polymers consisting of three to ten monosaccharide units joined together by glycosidic linkage.

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	3. () is a polymer of repeated glucose units linked together by β -glycosidic linkages.
	4. () is the most sugars that transported in phloem.
	5. () can be either a source or sink, depending on the plant's stage of development and
	the season.
B- W	rite short notes on: (3.5 marks)
	1. Steps of carbohydrates translocation within the plant.
	2. Effect of cytokinins on carbohydrates translocation.
	3. Role of companion cells in food translocation.
	4. Polysaccharides.
	5. Direction of carbohydrates translocation through the plant.
C- W	rite short notes on: (6 marks)
	1. Oleosomes.
*	2. Synthesis of fatty acid (equations only and don't forget the name of enzymes that activate the
	reactions).
	Section IV: Lipids – Amino acids & Proteins
A- Di	iscuss each of the following: (11 marks)
	1- Glyoxylate cycle. (3 marks)
	2- The conversion of 1-Acyl-glycerol-3-phosphate (lysophosphatidic acid) to diacylglycerol. (2 marks)
	3- Catabolism of an amino acid (alanine) by oxidative deamination method, referring to structural formula. (2 marks)
	4- Biosynthesis of glutamine from glutamate, referring to structural formula. (2 marks)
	5- The elongation in the translation step of protein synthesis process (Draw only). (2 marks)
B- Co	omplete the following sentences: (5.5 marks)
	1 & are the major types of lipids.
	2- Fructose-1,6-diphosphate is converted to fructose-6-phosphate by enzyme,while
	glucose-6-phosphate is converted to glucose by enzyme
	3- In cytosol, oxaloacetate is converted to
	conversion of lipids to sugars in germination. 4- A fatty acid composed of 18 carbon atoms is broken down to molecules of acetyl
	CoA.
	5is the final product of the conversion of lipids to sugars during germination.
	6 are used as the starting material for the β -oxidation of fatty acids, while
	is the initial acceptor for triacylglycerol biosynthesis.
	7 & are functions of RNA polymerase.
•)	8- The structural formula of the general equation of transamination reaction is
	9- The structural formula of aspartate-glycine-glycine tripeptide is
C	ampaya batwaan tha following in a tables (4.5 mayles)
C-C	ompare between the following in a table: (4.5 marks) 1- Codon and anticodon. (1.5 mark)
	2- Promoters and termination signals. (1.5marks)
	3- mRNA and rRNA (Function only). (1.5 marks)
	With our best wishes
	Prof. Sami Abo El-Kasem Dr. Rasha M. Eid Gamel
	Dr. Amany M. Kazamel Dr. Shimaa N. Tourky
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2. (.....) consists of glucose and galactose units.

Mansoura University
Faculty of Science

Physics Department

First term Exam, 3/1/2016

2nd level

Time allowed: 2 hours

Full mark: 80 marks

Subject: physics

Course: 221 Physical optics

Answer the following questions:

1- a) Give the optical arrangement to get Fraunhofer diffraction pattern using a rectangular single slit. Discuss this diffraction pattern. Drive the formula of intensity distribution of the resultant pattern.

(19 marks)

b) A grating with 6000 lines/cm is illuminated with monochromatic light at normal incident, the second order spectral line is observed to be deviated through 30°. Calculate the wavelength of the spectral line.

(8 marks)

2- a) Give a brief account, with an explanatory diagram of the optical arrangement of Fabry-Perot system of multiple- beams interference. Drive an expression for the intensity distribution in transmission for this system when the two coated plate are of same transmission coefficient T and of same reflectivity R. Sketch schematic diagram for the intensity distribution.

(20 marks)

b) Drive Malus law of the intensity of polarized light transmitted through analyzer.

(7marks)

3- a) Demonstrate an explanatory diagram of the optical arrangement of Young's double slits experiment. Drive the necessary formulae for the brightness and darkness conditions.

(10marks)

b) Give an experiment to determine the thickness of a thin sheet of transparent material using Fresnel's biprism. Drive the necessary formula.

(8 marks)

c) A water film (μ = 1.33) in air is 3000 A° thick if it is illuminated with white light at normal incidence. What color will appears to be in reflected light?

(8 marks)

Good Luck

Prof. Dr. Taha Sokkar



First Term Exam Level Two: CHEM 235 Time Allowed: 2 Hours Full Mark: 60 Marks Date: 10 January, 2016

Answer the following questions:

Question (1): (15 Marks)

(A) Complete the following reactions by the major product(s):

(a)
$$CH_3$$
- C - CH_3 + $3I_2$ aq. $NaOH$ + ...

(b) 2 — CHO + aq. $NaOH$ + ...

(c) CH_3 - CH - CH_2 - CH_3 + CI_2 — CH_3 + ...

(d) CH_3 - CH_2 - CHO + CH_2 - CH_3 + ...

(e) CH_3 - C = CH_2 — CH_3 — CH_3 - C = CH_2 — CH_3 - C = CH_3 — CH_3 - C = CH_3 — CH

(B)An alkene with the formula C_8H_{16} reacts with ozone (O₃/Zn-H₂O) to provide acetone and pentanal. What is the structure of this alkene?

Question (2):

For compounds from (1) to (4), answer the questions (a) to (g): (15 Marks)

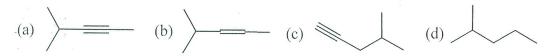
- (a) Give the IUPAC names for compounds (1), (2) and (3)?
- (b) How can you convert compound (1) into compound (2)?
- (c) Give the product when compound (1) is oxidized by potassium dichromate (K₂Cr₂O₇).
- (d) Suggest a suitable method to prepare compound (3).
- (e) How can convert compound (2) into compound (4).
- (f) Action of phosphorus trichloride (PCl₃) on compound (4).
- (g) Reduction of compound (1) with Zn-Hg in the presence of HCl.

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Question (3): (15 Marks)

(A)	Choose	the	correct	answer

(1) Which from the following chemical structures has the IUPAC name: 4-Methyl-2-pentyne



- (2) The IUPAC name of the alcohol CH₃CH(CH₃)CH₂CH(OH)CH₂CH₃ is:
 - (a) 2-Methyl-4-hexanol
- (b) 2-Methyl-4-hexanone
- (c) 5-Methyl-3-hexanol
- (d) 5-Methyl-3-hexanone
- (3) Hydration of CH_3 - $C\equiv CH$ with water and H₂SO₄/HgSO₄ produces:
 - (a) CH₃CH₂CHO
- (b) CH₃COCH₃
- (c) CH₃CH₂CH₃
- (d) CH₃CH=CH₂
- (4) Addition of two moles of HCl to 1-butyne produces?
 - (a) 1,1-dichlorobutane
- (b) 1,2-dichlorobutane
- (c) 2,2-dichlorobutane
- (d) 2,3-dichlorobutane
- - (a) quaternary
- (b) tertiary
- (c) secondary
- (d) primary
- (6) The major product that produced from the action of alc. KOH on CH₃CH(Cl)CH₂CH₃ is:
 - (a) CH₂=CHCH₂CH₃
- (b) CH₃CH=CHCH₃
- (c) CH₃CH(OH)CH₂CH₃
- (B) Suggest a suitable method to differentiate between 2-butanol and 3-pentanol.

Qustion (4): By equations only, Expalin the following reactions? (15 Marks)

- (1) Reaction of benzaldehyde with acetic anhydride in presence of CH₃COONa.
- (2) Conversion of 2-butene into 2-butyne
- (3) Heating of 2-methyl-2-butene with basic KMNO₄ solution.
- (4) Kolbe electrolysis KOOCCH2CH2COOK.
- (5) Reaction of acetaldehyde with two moles of methanol to form Acetal.

GOOD LUCK

Prof. Dr. Ahmed Fadda, Prof. Dr. Margret Mansour, Prof. Dr. Ehab Abdel-latif Dr. Manal El-fidawy, Dr. Ghada Emad and Dr. Ibrahim Youssef

Mansoura University

Faculty of Science

Chemistry Department

Course: Represented Elements

Code: Chem 221



First Semester

2nd Level Biochemistry

Date: Jan. 2016

Time: 2 hours

Marks: 80

		Answer The Following Questions	
	T	D. Comment on (Tombe) of the following: (29 montes)	
	I)		
	,	1) Thallous (I); Tl ⁺ compounds are stable.	0) :- 1
		2) The 1st Ionization Energy (1st IE) of (4Be, 7N and 10Ne) is high while for (8	
	3)	3) The reaction of elements of Group IA with water is increasing through the	
	4)	4) Berylium metal is amphoteric whereas aqueous solution of Be(II) is acidi	c.
	5)	5) Carbon monoxide is considered as good reducing agent.	
	,	6) Magnesium(II) chloride is heavily hydrated more than Barium(II) chlorid	
	7)	7) Nitrogen (N ₂)molecule is generally unreactive while Phosphorous molecu	
	,	8) The solubility of most of the salts of alkali group (IIA) elements is decrea	
	9)	9) The acidity and pka values in the hydrolysis of one molecule and the	ree molecules of boric acid (H ₃ BO ₃)
	10	10) Effect of increasing CO ₂ or O ₂ concentration on the blood pH.	
••••	••••		
		II) A- Write shortly on (4 only) of the following:-	<u>ks)</u>
		1) Biological importance of <u>carbon dioxide</u> .	
		2) Separation of <u>pure silicon element (Si) from silica ore (SiO₂)</u> .	
			y days.
•	4)	4) Isolation of <u>pure aluminium (Al) from bauxite ore {AlO(OH)}</u> .	
	5)	5) Production of nitric acid (HNO ₃) by Ostwald process.	
	6)	6) Bond strength of the pairs of (C-C & Si-Si) and (C-O & Si-O) bonds	
			5
II)		B-Complete (4 only) of the following chemical equations:-	3 marks)
		1) Ba + O ₂ (at 500 °C) \rightarrow 2) H ₂ BO ₃ + H ₂ O \rightarrow	
		3) Be ₂ C + H ₂ O \rightarrow 4) ¹⁴ N + ¹ n \rightarrow	
		7 0	A.
		5) $\text{Li}_3\text{N} + \text{D}_2\text{O} \rightarrow$ 6) $\text{B}_2\text{O}_3 + \text{NH}_4\text{BF}_4 \rightarrow$	
• • • • •			
III.	4.)		
		1-Discuss the structure and nature of bonding for	
		a) Diborane (B ₂ H ₆) b) Trimethylamine {N(CH ₃) ₃ } (7 mark	(2)
	a)	a) Diborate (B2116) b) Trimethylanine (14(C113)3)	3)
	2-/	2-An insulator like Silicon (Si) can be converted to semiconductors (n-type &	p-type) (5 marks)
II	IB)	(B) Choose the most correct answer for 8 only:- (12 marks)	
	,	1) Li + $O_2 \rightarrow \dots$, while Rb + $O_2 \rightarrow \dots$ a) Li & RbO b) Li O_2 & Rb O_2 c) Li ₂ O & Rb O_2 d) LiOH	H & RbO ₂
		2) $CaC_2 + N_2 \rightarrow \dots$ a) $Ca_3N_2 \qquad \qquad b) CaCN_2 + C \qquad c) CaCN_2 \qquad d$) C
			on M

) The structure of M	SII13/3 II as				
a) triagonal	b) triagonal	l pyramidal	c) due to sp ²	d) a & c are corre	ect
4)Diamond is	than grap	hite due to			
a) harder, saturation		2	c) weaker, sp ³	d) harder, sp ²	
5)Pb ²⁺ is stable than 1	Pb ⁴⁺ due to				
a) metallic character		ert pair effect	c) reactivity of	6s ² electrons d) b	& c
6) Baking powder (a) NaHCO3				ng baking) CaHPO4	
7) $H_2SO_4 + SO_3 \rightarrow$ a) fuming sulphuric e) b & c		b) H ₂ S ₂ O ₇	c) H ₂ SO ₃	d) a & b	
8) Oxidation state of	Cl in HClO ₄	is			
a) +1	b) +7	c) +5	d) +4		
9) ⁷ Li isotop is used i	in treatment (of cancer via			
a) neutron capture	therapy	b) physiother	apy c) chemother	apy d) electron cap	ture therapy
10) Chlorophyll is a) Ca ²⁺ , gypsum				process rage d) Mg ²⁺ , pl	ıotosynthesis
11)is used as	anti-acid for	ulcer patients			
a) CaSO ₄		c) BaSe	O ₄ d) NaHo	CO_3	

Best wishes

Prof. Tawfik Rakha

Prof. Sahar Mostafa

Dr. Rania Ramadan

Mansoura University
Faculty of Science
Botany Department
El-Mansoura, Egypt



جامعة المنصورة كلية العلوم قسم النبات المنصورة - مصر

Final Examination in Botany Second Term: Jan. 2016

Educational Year: Second Level Program (Branch): Biology

Subject: Bot (201) Course(s): Introduction to Plant Ecology & Taxonomy

Time: 2 hrs Date: 17 / 1 /2016 Full mark: 60 Question

Question mark: 20

Answer the Following Questions:

Q.1: [A] Mark the following sentences by true ($\sqrt{\ }$) or false (x) (10 marks)

- 1) Drought escaping plants are short lived plants called ephemerals.
- 2) Xerophytes adapted to the drought by developing the root system.
- 3) A vertical section of soil through all horizon termed as soil profile.
- 4) The leaves of some xerophytes rolled to cope with the dry habitat.
- 5) Prismatic soil structure is commonly found in soil of arid and semiarid region.
- 6) Hydrophytes adapted to habitat by formation of thick cuticle.
- 7) Mangrove belongs to xerophytic vegetation.
- 8) Physical drought means that, the water is present in excess amount but it is not available to plants.
- 9) Hydrophytes contained well developed conducting tissue.
- 10) Amphibious hydrophyes, these plants are adapted to both aquatic and terrestrial mode of life.

O.2: [A] Write on each of the following:

1- Evolution of vegetation.

(5 marks)

2- Xerosere succession. (please draw)

(5 marks)

10. The soil parents materials may be or

	1- In wind-pollination, pollen grains are characterized by . and	(10 marks)
	2 and constitute the essential leaves of the f	lower.
	3- Theophrastus classified the plants according to their for, and	rms into
	4- Pistil is composed of, and	
	5- Imbricate aestivation may be or	hey may be
	7- In parietal placentation the number of carples are number of lobules are	. while the
	8- Collectively the stamens are termed	
Q.3:	[A] Describe and draw each of the following:	
Q.5.	1- Simple dry dehiscent fruits.	(3 marks)
	2- Racemose inflorescences with sessile flowers.	(3 marks)
	3- Floral diagram of <i>Phoenix</i> flowers.	(4 marks)
	[B] Mark the following sentences by $\sqrt{\text{ or X}}$ (with corre	ection)
		(10 marks)
	1- Androcium in family: Compositae are diadelphous.2- Stamens with united filaments in family: Cruciferae.	
	3- The anthers are united with free filaments in Geranina	ceae.
	4- Sympetalae are characterized by free petals.	
*	5- Malvaceae have hypogenous flowers.	

With Best Wishes & Good Luck

Examiners:

Prof. Dr. Ibrahim A. Mashaly Prof. Dr. Mohamed E. Abu-Ziada

Mansoura University Faculty of Science Physics Departement Subject: Introduction to Biophysics

Course code: biophys221



First Term, Final Exam 2nd Students Time Allowed: 2 h. Date: 20/1/2016

Full Mark: 80 Mark

Answer all the following questions

I-Write short notes about each of the fo
--

- a) Cobalt 60 and linear accelerator
- c) Classification of light atom interaction
- b) Ion distribution in cell membrane
- e) Acoustic impedance
- f) Transducer
- g) Equivalent Circuit Model for the Plasma Membrane
- h) Treatment planning software
- d) Compton Effect

II-Choose the correct answer from the followings:

- 1) The process in which α and β rays pass close to atoms and knocks the electrons out is called:
- a) Ionization

b) Ionisation

b) Decay

- d) None of above
- 2) The sound that emanates from a piezoelectric transducer originates:
- a) From a point on the active surface
- b) From most of the active surface
- c) From a small area in the center of the active surface
- d) From the edges of the active surface
- 3) Period is determined by:
- a) Sound source

b) Medium

- c) Both
- 4) The time it takes a wave to vibrate a single cycle, or time from the start of a cycle to the start of the next cycle:

a) Period	b) Frequency
c) Wavelength	d) Speed
e) Power	
5) Which of the following ions	s are involved in neuronal action potentials?
a) Na ⁺	b) K ⁺
c)Cl	d) A and B only
e) A, B, and C	
6) At what membrane voltage activated	ge do neuronal voltage-gated Na+ channels become
a) -70 mV	b) -55 mV
c) 0 mV	d) +55 mV
7) At what membrane voltagactivated?	ge do neuronal voltage-gated K+ channels become
a) -70 mV	b) -55 mV
c) 0 mV	d) -90 mV
8) The hyperpolarization pha	se of the action potential:
a) Is due to the opening of vol	tage-gated Cl- channels
b) Is due to the prolonged ope	ening of voltage-gated K+ channels
c) Is due to the closure of rest	ing Na+ channels
d) None of the above	
9) What is a major health con	cern wth MRI?
a) Reaction to applied drug	b) extrerme cold?
c) Radiation dose	d) localized burns due to metallic implants?
10) Uses high doses of radiation precisely to avoid damaging h	on to kill cancer cells and shrink tumors, delivered ealthy brain tissue.
a) Radiation therapy	b) Ionizing radiation
c) X-ray	d) Radiosurgery
11) Which of the following is	NOT true about the neuronal action potential?

a) Action potentials are all-or-nothing.		
b) Action potentials travel along axons in a non-decremental fashion.		
c) Repolarization and hyperpolarization are due to the activity of K+ channels.		
d) All of the above are true about action potentials.		
12) Which of the following is NOT a source of background radiation?		
a) Radiation from Naturally occurring unstable isotopes.		
b) Radiation from a Source being measured.		
c) Radiation from Space.		
d) Radiation from Human Activity.		
13) Which of the following types of radiation can enter living cells and cause ionization, thus damaging or destroying the cell?		
a) Gamma. b) Alpha and Beta.		
c) Beta and Gamma. d) Alpha, Beta and Gamma.		
14) Where does radiation come from?		
a) An electron b) An atom.		
c) A stable nucleus d) An unstable nucleus which decays.		
15) Which type of radiation would be stopped by a few millimetres of aluminium, but not by paper?		
a) Gamma. b) Infra-red.		
c) Alpha d) Beta.		
III-Write the scientific expression:		
a) The component of the ultrasound imaging equipment that is placed in direct contact with the patient's body().		
b) Conversion of electrical energy to mechanical energy and vice versa (
c) Nerves that communicate messages between the central nervous system and the rest of the body nerves that communicate messages between the central nervous system and the rest of the body().		

- d) Places radioactive material into tumor or surrounding tissue(
- e) The action potential goes past -70 mV because the potassium channels stay open a bit too long().
- f) A pair of reflecting surface of which one is a perfect reflector and the other is a partial reflector(

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مع تمنياتي بالتوفيق د/أمل الشهاوي



Mansoura University **Faculty of Science Zoology Department**

Academic year: 2nd Year Program: All Programs

Code: 204 Z

Course: Chordates and Vertebrates

Marks: 60

Date: SUN, 24/1/2015

Time: 2 Hours

Number of papers: 2

Answer ALL the following questions:	
Question One:	[30 Marks]
[A]- Choose correct Answer (ONLY one answer is correct)	[10 Marks
1) - In which of the following animals the heart does not have the left and right auricle	207
(a) Lizards and snakes (b) Frogs and toads (c) Cartilaginous and bony fishes	
2) – Which of the following structures is present in all chordates?	(d) Crocodiles and alligate
(a) Cranium (b) Notochord (c) Spinal cord	(d) Wantal 1
3) – Egg-laying mammals are of the subclass	(d) Vertebral column
(a) Marsupials (b) Placental (c) Eutheria	
4) – All of the following involved in swimming of Osteichthyes except	(d) Monotremes
(a) Homocercal cau ial fin (b) Squelence (c) Bilaterally body compression	(1) 5
5) - The infundibular region of Amphioxus lanceolatus has pit, which is a	(d) Swimming bladder
chemoreception.	esponsible for olfaction an
(a) Optic (b) Lorenzini (c) Auditory	(d) Kolliker's
[B]- Complete the following sentences.	[10 Marks]
6) - The first 1 st visceral arch of splanchnocranium of Scyliorhinus canicula is c	
composed of and	alled and it is mainly
7) – The pigments which are present in the skin of toad are, and	
8) – The sensory organs of Petromyzon fluviatilis are, and	
9) – The exoskeleton of Chondrichthyes is represented by which is composed of -	
10) – The skin of birds is dry, cornified and devoid of glands exceptgland.	,, and
[C]- State if the sentence is true (T) or false (F), and postulate a correction if possib	In [10 May 1]
. Correction in possib	le. [10 Marks]

11) - Tunicates have a notochord at some point in their life cycle.

[10 Marks]

- 12) Skull of reptiles are dicondylic.
- 13) Amphibians have 4-chambered hearts.
- 14) Fertilization in Branchiostoma lanceolatus is external and gives rise to Ammocoet larva.
- 15) Salamander and Newt are typical examples of Anura.

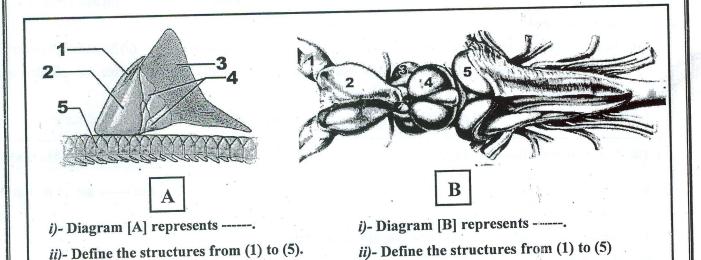
The rest of questions in the next page \downarrow

Question Two: Write a brief notes on the following items: [15 Marks] [4 Marks] 1) - State (6) main general characters of class Mammalia. 2) - Male & urinogenital system of Scyliorhinus canicula. (Draw) [4 Marks] [4 Marks] 3) - Digestive system of *Petromyzon fluviatilis*. (Draw) [3 Marks] 4) - Air sacs of birds. Question Three: Answer the following items: [15 Marks] 1) - State (5) features of flight adaptations of various body organs of birds. [4 Marks] 2) - write short notes about the swimming bladder structure and its function of boney fishes. (Draw) [3 Marks]

3) - Compare between Lacertilia and Ophidia.

ii)- Define the structures from (1) to (5).

4) - Answer the questions concerning the following two diagrams [A] and [B]:



Good Luck

Examiner: Dr. Ahmed Elmansi

[4 Marks]

[4 Marks]

Mansoura University
Faculty of Science
Zoology Department



First term, Final exam, January, 2016

Education Year: Second Level Date: 27\ 1\ 2016 Time: 2 hours Cod: Z201	Program: All programs of biology Subject: Zoology Courses: Introduction to Embryology Full Mark: 60
Answer all the	following question:
Q1- A) Rewrite 14 sentences only in your a	nswer sheet after correction: (14) marks
1- The female puberty-hood starts under the a- Hypothalamus b- Pituita	ry gland c- both of them respectively
2- The man who has sperm dea a-necrospermia b- oligosperm	d sperms in the ejaculate. nia c- aspermia
3- Sperm capacitation is associated with a- adding to b- removal of	adherent seminal plasma protein. c- union with
4- LH stimulate the production of	
5- The bulbourethral secretion gives the sen a-milky appearance b- acidi	ty c- alkalinity
6- At the puberty, the PGCs differentiate int a-spermatogonia b-Sertoli ce	o which are spermatogenic linage. ells c- spermatids
7- The egg of frog is classified asa- a- alecithal b- mesolecit	according to the amount of yolk. hal c- heavy telolecithal
8- During spermeiogenesis, Spermatids will a- with mitotic b- with meio	
9- The activation of SRY gene on di gonads. a- XY chromosome	fferentiates the indifferent embryonic gonads into male b- Y chromosome c- XX chromosome
10- Sperm secretes Hyaluronidase that helps a- corona radiata b- zona pelluc	to penetrate the ida c- plasma membrane
11- Secretion of prostate gland contains fibri a- liquefies b- solidifies	nolysin enzyme that the semen. c- neutralizes
12- The empty Graffian follicle becomes a- primary oocyte b- corpus luteu	

13- PGCs migrate by amoeboid movement from the	
14- During spermeiogenesis, shedding of most of the cytoplasm occurs during a- cap-phase b- acrosomal-phase c- maturation-phase	••••
15 - The amount of yolk influence the	
a- egg size b- pattern of cleavage c- both of them	
Q1-B) Discuss the important of the following phenomenon:	(6) marks
1- Egg activation and the cortical reaction of fertilization.	2 g
2- Post-fertilization events.	
Q2-A) Compare between TWO pairs only of the following:	(10) marks
1- Spermatogenesis and oogennesis process.	
2- Somatic cells and sex cells.	
3- Proliferative and secretory phase of the uterine cycle.	
Q2-B) Compare between the steps of gastrulation in both amphioxus and toad, ad diagrams.	ding labeled (10) marks
Q3-A) Write short note on each of the following referring to its function:	(6) marks
1- Allantois 2- Yolk sac 3- Cleavage	
Q3-B) Write briefly on the following:	
a- Types of monozygotic twins.	(4) marks
b- Steps of implantation of human embryo, adding labeled diagram.	(10) marks
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