University of Mansoura
Faculty of Science
Department of Zoology
Educational year: Fourth Level
Program: Chemistry/Zoology



Course: Environmental Pollution and

Biological Analysis (Z 405) (Opt.)

Date: 2 January 2016

Time: 2 hours
Full Mark: 80 Marks

Q_1 - Answer as true (\checkmark) or false (X)

[20] Marks

- 1. Nanotechnology is research and development to control or manipulate matter on molecular or atomic scales, with lengths between $1-100~\mu m$
- 2. Electrophoresis of positively charged particles is called anaphoresis
- 3. Biomes are all possible ecosystems
- 4. To obtain crude nuclei, the cells must be centrifuged at 700 g for 10 min
- 5. Separating funnel is used in liquid-liquid extraction technique
- 6. Many toxicity tests examine specific types of adverse effects, known as "endpoints"
- 7. Power stations are examples of the diffuse pollution
- 8. $1 \text{ mg/kg} = 1 \mu\text{g/ml} = 1 \text{ ppm}$
- 9. Nanorobots moves with blood stream and guided with laser
- 10. Noise pollution affects nesting of songbirds

Q_2 - C	Choose the correct answer:		[20] Marks
1.	A hazy layer called	may hang over industrial citi	es indicates to air pollution
	a) smog	b) fog	c) smut
2.	is toxic symptotime frame < ½ the life- time o		
	a) chronic toxicity	b) subchronic toxicity	c) acute toxicity
3.	is a technique of	instrumentation	
	a) protein precipitation	b) centrifugation	c) spectrophotometry
4.	Survival and reproduction are	biomarkers at the	organization level
	a) cells	b) individual	c) community
5.	Ecosystem is a core concept in	n and Ecology	
	a) hiology	b) geology	c) histology

	6.	All ecosystems are re	equiring a net flow of energ	y to persist.	
		a) open systems	b) closed systems	c) both of them	
8	7.	is a tool may detect a	and treat the cancer in a ver	y small amount of	cells
		a) nano-engineering	b) nano-device	c) nano-science	
	8.	In the planar chromatography, the	e stationary phase is	••••••	
		a) glass	b) aluminum sheet	c) a fine powder	
	9.	confirm diagnosis of	f acute or chronic disease as	cancer tumors	
		a) bioassays	b) biomarkers	c) bioindicators	
	10	deals with subatomic dynamic processes in matter with			ons to probe
		a) Attoscience	b) Femtoscience	c) Nanoscience	
23	3- C	hoose the alien word(s)			[30] Marks
	1.	spectrophotometry – electrophore	esis – centrifugation – chron	natography	
	2.	$10^9 \text{ am} - 10^6 \text{ fm} - 1000 \text{ pm} - 1000$	0 nm – 10 μm		
	3.	water – consumer – primary prod	lucer – energy – decompose	r	
	4.	fitness – biomarker – biodiversity	y – bioassay – animal densit	У	
	5.	shipping – dust of solid rocket m	otors – space missions – sat	ellites – aircrafts	
	6.	$LD_{50} - LC_{50}$ - mortality - dose -	test animals .		
	7.	pesticides – biodegradable – dom	nestic wastes – sewage – pla	stics	
	8.	zone plate – nanotechnology – bi	iosensors – virus – 10 ⁻¹⁵ sec	ond	
	9.	biosphere – populations – organ	s – Earth – biomes		
	10). diarrhea – cough – water – typho	oid – pollution		
Q	₄ - C	Compare between both of:			[10] Marks
	1.	Liquid chromatography and gas	chromatography		
	2.	Effects of air pollution and water	r pollution on human health		

" BEST WISHES "
Dr. Waleed Khaled Elaidy

Mansoura University
Faculty of Science
Chemistry Department
Subject code: Chem. 415
Course: Electro-analytical
chemistry and spectroscopic
methods of analysis



First semester examination 4th level students Program: Chemistry/Zoology

and Chemistry/Botany

Date: 12/1/2016 Time allowed: 2 hours Full mark: 80 marks

(الأسئله في صفحتين) :Answer the following questions

Section A: (Spectroscopic methods of analysis) (40 marks)

Question 1: (20 marks)

- a. Define each of the following: (10 marks)
 - 1. Electromagnetic radiation.
 - 2. Wave length.
 - 3. Scattering.
 - 4. Auxochromes.
 - 5. I.R active substances.
- b. Draw the diagram which represents: single beam instrument. (5 marks)
- c. <u>Calculate the molar absorptivity of</u>: K₂Cr₂O₇ at 455 nm, given that 36.5 mg was dissolved in 500 mL exhibits 12% transmittance at 455 nm in a 2-cm cell. (5 marks)

Question 1: (20 marks)

- a. Put true ($\sqrt{ }$) or false (\times) and correct the wrong one: (10 marks)
 - 1. According to Beer's law, A α 1/C.
 - 2. During the interaction between EM radiation and matter, the type of excitation independent on the wave length of the light.
 - 3. Wagging vibration involves change in bond length.
 - **4.** Spectroscopic methods are not classified according to the region of the electromagnetic spectrum.
 - 5. The colour in KMnO₄ is due to defect in crystal.

b. Complete the following: (10 marks)

- 1.lamp is a source for I.R radiation while..... lamp is a source for U.V radiation.
- 2.is an example for I.R detector.
- **3.** Electrons are promoted to higher orbitals by......while vibrations are excited by......and rotations are excited by......
- 4. Bathochromic shift is a shift to and it also called
- 5. Hyposo-chromic shift is a shift to.....and it also called

Please turn the page →

Section B: (Electro-analytical chemistry) (40 marks)

Question 3: (20 marks)

- a. Complete each of the following sentences: (5 marks)
 - 1. In static method......passes through the electrochemical cell like.....
 - 2. Electrodes of the fourth kind can follow the reaction mechanism for
 - 3. Coulometry requires 100% current efficiency which means:
 - 4. For CPC technique, two types of working electrodes are commonly used:.....or......
 - 5. Irreversible peak current is proportional to:.....and......and......and......
- b. Put true ($\sqrt{\ }$) or false (\times) and correct the wrong one: (5 marks)
 - 1. Galvanostat is a device used to control the current in the dynamic methods.
 - 2. Potentiometric end point of acid-base titration is determined using glass electrode.
 - 3. Fluoride ion electrode interferes by **OH** at high pH values.
 - 4. During coulometric titration of Fe²⁺ against cerium, Ce³⁺ is the mediator.
 - 5. Mass transfer due to convection can be eliminated by preventing stirring process.
- c. Sketch the diagram which represents: the cell used for polarographic measurment. (4 marks)
- d. The purity of a sample of picric acid, C₆H₃N₃O₇ (M.wt.=229 g/mole), is determined by controlled-potential coulometry, converting the picric acid to triaminophenol, C₆H₉N₃O. A 0.03 g sample of picric acid is placed in a 1000-mL volumetric flask and diluted to the volume. A 10-mL portion of this solution is transferred to a coulometric cell and diluted till the Pt cathode is immersed. The exhaustive electrolysis of the sample requires 1737 C of charge. Report the purity of the picric acid. (6 marks)

Question 4: (20 marks)

- a. Give the scientific name for each of the following statements: (5 marks)
 - 1. It completes the electric circuit and its potential known and remains constant.
 - 2. It contains a thin glass bulb at the bottom that is selective to \mathbf{H}^{+} ions.
 - 3. It contains quantitative deposition of analyte as a solid on the cathode surface.
 - 4. It can be removed by addition of Triton X-100 to the polarographic cell.
 - 5. The most widely used technique for quantitative analysis of redox reactions.
- b. Mention two advantages for each of the following: (4 marks)
 - 1. ISE's.
 - 2. CCC.
- c. Comment on the following: (5 marks)
 - 1. The measured pH by glass electrode is higher than the actual pH in highly acidic solutions.
 - 2. DME is not very useful for oxidation processes.
- d. Vitamin C, $C_6H_8O_6$ (M.wt.=176 g/mole), gives an anodic wave at +0.1 V vs SCE that can be used for analysis. 20 mL sample of orange juice stabilized with the addition of a small amount of oxalic acid is filtered, buffered to pH 8 and diluted to 50 mL. The anodic wave has I_d value of 9.36 μ A with drop factor ($m^{2/3}t^{1/6}$) value of 2.6 and the diffusion current constant ($I_d/m^{2/3}t^{1/6}$ C) equal 3. Calculate vitamin C concentration in ppm unit. (6 marks)

Good luck: Dr. Yasmeen Gaber and Dr. Hany Moustafa



Mansoura University Faculty of Science Zoology Department Subject: Zoology 401

First Term

Fourth Level: Chemistry & Zoology

Date: 16-1- 2016 Time Allowed: 2hours

Courses (Comparative anatomy and origin of species)

Answer the following questions (Full marks 60)

Q1 - Fill in spaces with correct words (30	Marks)
 a- The methods of coelom formation are	odified to form4 which consists in portion named6 which coated by re9,10,11,12- and
e- The ear ossicles of the middle ear compoderived from21,22and2324 only. f- The stomach of birds composed of -25 composed of27,28,29 and	respectively and characteristic to class and26 while in ruminants it
Q2- Choose the correct answer:	(10 marks)
 Reduced coelom cavity found in a-nematods & arthropoda. c- arthropoda only. Otic bones presented only by periotic in classes. 	b- annelida. d- nematode only.
a- amphibian.	b- aves.
c - reptelia .	d- mammals.
3- The secondary palate of crocodile isa-floor of neurocranium.	b- totally fleshy.
c- anterior hard & posterior fleshy.	d- totally bone.

4- Double circulation characteristic to -----

a-mammals only.	b- all hot blooded animals.
c- all tetrapods.	d- all amniotes.
5. According to the cuene form	, the herbivorous mammals have not the following
shape	, the herbivorous mammais have not the following
a-selenodont.	b- lophodont.
c- secodont.	d- lophodont & selenodont.
6- The provertebra appeared in	n
a-cephalochordata.	b- bony fishes only.
c- cyclostomata & aves.	d- cyclostomata only.
7-The whole body skin devoid o	f exoskeleton in
a-cephalchordata only.	b- cyclostomata only.
c- cephalchordata & cyclostoma	ta. d- all aquatic animals.
8-Jaw suspension in the amphi	stylic type occurred by
a-ligaments only.	b-hyomandibular only.
c- hyomandibular &2processes.	d-4processes from upper jaw itself.
9- Pleuro protineal cavity prese	ent in
a-mammals only.	b-amphipia only.
c- birds&hipia.	d-reptilia only.
10- Hooves are characteristic to)
a-primates only.	b-ungulates only.
c- primates & ungulates.	d-all mammals.
- Draw with complete label	ing the following items (20 marks)
a- The different types of reptelb- Phylogeny of the heart cham	ian skull. ibers among different tetrapoda.

(With best wishes Dr: Fawkeia Elsayad)

d- The respiratory system of both cyclostomes and cartilagenous fishes.

c- Different shapes of the articulating surfaces of vertebrea.

Q3

Mansoura University **Faculty of Science Chemistry Department**

Subject: Chemistry

Course(s): Surface Chemistry and

Molecular Spectroscopy

Code: Chem. 445



First term

4th year level Chemistry-Zoology & Chemistry-Botany Students Date: 23 January 2015

Time allowed: 3 hours Full mark: 60 marks

Answer the following questions:

_	
	Section A: Molecular Spectroscopy
Q1	: Complete the following: [12 Marks]
1.	The energy of light is directly proportional to
2.	Atomic absorption is, while atomic emission is
3.	The wave lengths of lines in the hydrogen emission spectrum is given by
4.	The total energy of molecule can have is equal
5.	The intensities of spectra line in rotational spectrum are directly proportional to
6.	The molecules are infra-red active they must have, while the molecules
	are Raman active they must have
7.	Chromophore is and the auxochrome is
8.	The possible electronic transition for organic compounds are
9.	The number of vibration for linear molecules areand non-linear
	molecules are
10.	The vibrational spectra are accompanied bywhile the electronic
	spectra by
11.	When molecule absorbs infra-red radiation the molecule isand
	when molecule absorbs microwave radiation the molecule is
12.	The molecule contains a double bond has force constantthan the
	molecule contains a single bond
Q2	: Explain the following: [10 Marks]
1.	The difference between IR and Raman spectra.
2.	Rayleigh, Stokes and anti-Stokes scattering.
3.	The modes of vibration of CO ₂ and H ₂ O
4.	Rigid rotator
5.	The electronic transitions for β -alanine: H_2N OH
03	: Calculate the following: [8 Marks]
	Calculate the energy per mole associated with ultra-violet region of electromagnetic
1.	Survey and and all has more appropriated that a trainer to dront of the controlled

- spectrum assuming that the wave length is $5x10^{-6}$ cm.
- 2. Calculate the bond length for $C^{12}O^{16}$ molecule knowing that moment of inertia is 6.2 x 10⁻⁴⁹ kg m².

Section B

Answer five questions only of the following, [6 marks for each question= 30 marks]

1) **Define** the following terms with aid of diagrams:

i. Interface

ii. Adhesion forces

iii. Critical temperature

iv. Wetting agents

v. contact angle

vi. Capillarity

- 2) **Determine** the spreading coefficient of n-hexadecane ($\gamma_{O1A}=30.0$; $\gamma_{O1W}=52.1$), n-octane ($\gamma_{O2A}=21.8$ $\gamma_{O2W}=50.8$), and n-octanol ($\gamma_{O3A}=27.5$; $\gamma_{O3W}=8.5$) on water ($\gamma_{WA}=72.8$) at 20°C. (*Comment*,,,,)
- 3) **Describe** the differences between physical adsorption and chemical adsorption.
- 4) **Deduce** the set equations of Langmiur model for describing the solid/gas adsorption.
- 5) Deduce and discuss the Kelvin Equation.
- 6) In the Du Nouy tensiometer, if the diameter of the ring is 1.0 cm and the force needed to pull the ring up (with the liquid attached to the outer and inner periphery of the ring) is 6.77 mN, what is the surface tension of the liquid?
- 7) Calculate the surface excess concentration for a 1 M aqueous solution of NH₄NO₃. The rate of change the surface tension with concentration equal unity. (R = 8.314 J/mol.K.)
- 8) Discuss the effect of temperature on surface tension of different liquids.

Best Wishes;

Prof. Essam Arafa,

Prof. Salem Samra,

Dr. Hamdy Farag,

Dr. Kamal Shalabi



Educational year: First Term Mansoura University Fourth Year Year: Faculty of Science 26/1/2016 Date: **Zoology Department** Program: Chemistry & Zoology Time Allowed: 2hrs Full Mark: 60 Marks Subject: Z 403 Course(s): Experimental Embryology Final Exam

30 Marks Attempt all the following questions

Q.I: Part 1)- MCQs:

Marks

(A) Please select the single best answer for each of the following questions:

1- Sertoli cells serve a number of functions during spermatogenesis? a.Secrete progesterone b. Secrete testosterone Phagocytosis

2-All of the following are requirements for cell to grow in vitro **EXCEPT:**

a-Tempreture at 37°C

b.Humidity

c.Growth medium

Selenium

b. Immortal

3-Advantages of embryonic stem cells are the following EXCEPT c. Hard to culture

d.Can be divided

a.Flexible many times

4-Which type of cloning involves separating cells from the same embryo to produce multiple offspring?

a. Natural cloning b. Fusion cell cloning c. Embryo transplants d.Genetic cloning

5-Classification based on level of stem cell differentiation

a. Totipotent

b.Megapotent

c.Protopotent

d Non

of them

Part II: True & False questions

Marks

1- Primary cultures derived from normal animal tissue.

True or

False

2- Alzheimer's disease and can cloning help?

True or

3-The causes of artificial insemination are the abnormalities preventing deposition of sperm close to the cervix

True or False

4-In genetic cloning, viral vector carrying healthy gene and inserts into cell and then new gene appear in the cell True or False 5-Embryonic Stem Cells are never derived from eggs fertilized inside of a woman's body. True or False Q.II: A) Fill in the spaces 5 Marks
1- Characteristics of continuous cell lines 1,2,3,2-Features of stem cells are 1,2,3,3. 3-In vitro fertilization and embryo transfer can be management infertility of couples suffering from 1,2,3,3,
3, 7-Sources of embryonic stem cells are 1,2, Q.II: B) Short Answer 15 Marks (1) Discuss Three only from the following: 1-Applications of Animal Cloning 2-How stem cells can cure baldness & diabetes 3-Micro fertilization of human egg 4-Main types of cell culture according to the cell morphology 5-Types of stem cells according to the donors? Part II: 30 Marks
Q1. Write short accounts of five of the followings 20 Marks:
 Regeneration of tail and lens of amphibian. Cell mediated immunity. How to construct fate maps of mammals. Renal and central nervous abnormalities. Numerical & structural chromosomal abnormalities. Limb abnormalities.
Q2. What do mean by only five of the following terms and their basis
10 marks: Antigen-antibody reaction, Spinna bifida occulata, gonadal
dysgenesis, Cranioschisis, Lobster claw, Club foot.
Best Luck
Prof. Dr. Hassan El-Sayyad Prof. Dr. Amoura Abou-El-Naga